In the Claims

Canceled Claims

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Claims Please cancel claims 12-17 and 18-20 as being drawn to non-elected groups II and III. Applicant expressly reserves the right to file divisional applications directed at the non-elected subject matter.

Current Status of Claims

1.(currently amended) A method of prevention sulfidation of metals comprising the steps of:

adding to a fluid including a sulfiding compound an effective amount of a preventative composition, where the composition reduces or prevents sulfidation by deactivating metal sites involved in the formation of atomic sulfur and/or sulfides at or on a surface of the metal and where the effective amount of the preventative composition is between about 0.2 ppm and about 0.8 ppm.

2.(currently amended) A method of stopping sulfidation of metals comprising the steps of:

adding to a fluid including a sulfiding compound an effective amount of a preventative composition, where the composition stops or arrests further sulfidation of the metal by deactivating metal sites involved in the formation of atomic sulfur and/or sulfides at or on a surface of the metal and where the effective amount of the preventative composition is between about 0.2 ppm and about 0.8 ppm.

- 3.(withdrawn)
- 1 4.(withdrawn)
- 5.(currently amended) The method of claims 1-2, wherein the composition comprises a compound having a higher affinity for the metal surface than the sulfiding compound.

- 1 6.(currently amended) The method of <u>claims 1-2</u>, wherein the composition comprises
- 2 an effective amount of a phosphorus in the form of a phosphorus-containing compound to
- 3 reduce sulfidation of the metal.
- 1 7.(canceled)
- 1 8.(canceled)
- 1 9.(canceled)
- 1 10.(canceled)
- 1 11.(canceled)
- 1 12.(canceled)

13.(currently amended)

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compound comprises phosphorus, phosphines of formulas PH₃, PRH₂, PR₂H, and R₃P where
each R is the same or different and is a C1 to C20 carbon-containing group including alkyl,
aryl, alkaryl or aralkyl; phosphites including ammonium phosphites; alkali metal phosphites;
alkaline metal phosphites; phosphites having organic counter ions; phosphates including
ammonium phosphates; alkali metal phosphates; alkaline metal phosphates; phosphates
having organic counter ions; pyrophosphates including ammonium pyrophosphates; alkali
metal pyrophosphates; alkaline metal pyrophosphates; pyrophosphates having organic

The method of claim $\frac{8}{6}$, wherein the phosphorus-containing

polyphosphates; alkaline metal polyphosphates; polyphosphates having organic counter ions;

counter ions; polyphosphates including ammonium polyphosphates; alkali metal

thiophosphates; thiophoshites; or other phosphorus-containing compounds capable of

12	inhibiting sulfuric corrosion of metal surfaces, or mixtures or combinations thereof.
1	14.(original) A method of pre-treating metal surfaces comprising the steps of:
2	contacting a metal surface with an effective amount of a pre-treating composition
3	sufficient to deposit onto the metal surface a protective coating, where the coating prevents
4	or reduces sulfidation of the metal by deactivating metal sites involved in the formation of
5	atomic sulfur and/or sulfides at or on the surface, where the effective amount of the
6	preventative composition is between about 0.2 ppm and about 0.8 ppm.
1	15.(original) The method claim 14, wherein the pre-treating composition comprises an
2	organo-phosphorus compound and the method further comprising the step of:
3	oxidizing the organo-phosphorus compound to a phosphorus oxide compound.
1 2	16.(original) The method claim 14, wherein the composition comprises a compound having a higher affinity for the metal surface than the sulfiding compound.
1 2	17.(original) The method claims 14, wherein the composition comprises an effective amount of phosphorus in the form of a phosphorus-containing compound.
1 2	18.(canceled) The method claims 14, wherein the effective amount of the phosphorus is between about 0.1 ppm and about 5 ppm in the fluid.
1	19.(canceled) The method claims 14, wherein the effective amount of the phosphorus
2	is between about 0.2 ppm and about 0.8 ppm.
1	20.(currently amended) The method claims 14, wherein the phosphorus-containing
2	compound comprises phosphorus, phosphines of formulas PH ₃ , PRH ₂ , PR ₂ H, and R ₃ P where
3	each R is the same or different and is a C1 to C20 carbon-containing group including alkyl

aryl, alkaryl or aralkyl; phosphites including ammonium phosphites; alkali metal phosphites; alkaline metal phosphites; phosphites having organic counter ions; phosphates including ammonium phosphates; alkali metal phosphates; alkaline metal phosphates; phosphates having organic counter ions; pyrophosphates including ammonium pyrophosphates; alkali metal pyrophosphates; alkaline metal pyrophosphates; pyrophosphates having organic counter ions; polyphosphates including ammonium polyphosphates; alkali metal polyphosphates; alkaline metal polyphosphates; polyphosphates having organic counter ions; thiophoshites; or other phosphorus-containing compounds capable of inhibiting sulfuric corrosion of metal surfaces, or mixtures or combinations thereof.

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